

# Effect of Aerosil and Oleic Acid on Sedimentation of Magnetorheological Fluid

HirenPrajapati, Jas Shahanand ,HitarthNimkar

**Abstract**—Magnetorheological Fluids (MRFs) are considered as smart fluids because they control viscosity using external magnetic field. It contains ferro-magnetic powder which are aligned in magnetic flux lines. The magnetic force between particles are controlled by magnetic field intensity. This controllable viscosity makes them acceptable in many mechanical applications, but due to difference in density between suspended particles and carrier fluid sedimentation is bound to occur. This thus creates the need of some additives. In our study, silica Nano particles (commercially known as Aerosil 200) is used as stabilizer and Oleic Acid is used as surfactant and their effect on sedimentation is studied in this article. Some other synthesis parameters like particle concentration, stirring duration and material loading also cause some change in sedimentation rate.

**Keywords:** Magnetorheological Fluid, settling, Synthesis, sedimentation.

## I. INTRODUCTION

Magnetorheological fluids (MRFs) are a class of smart materials. MRF consists micro sized iron particles dispersed in viscous carrier fluid.[1] These dispersed iron particles makes chain like structures in presence of magnetic field and converts into semisolids from liquid state. The formed particle-chains restrict fluid movement and increase yield strength of MR fluid. After removing magnetic field the semisolid form converts into fluid state back within milliseconds again. [2] Total viscosity (or Dynamic Yield Stress) of MRF is the viscosity in the presence of magnetic field and also the viscosity of MRF in absence of the magnetic field. Therefore MRFs are considered as non-Newtonian fluids. This rapid change of MRF viscosity in presence of magnetic field makes them suitable in lot many applications like isolators, shock absorbers, clutches, engine mounts, alternators, power steering pumps, control valves, brakes, dampers etc.[3] The operating range of said applications can be changed using MRF. The stability of iron particles, within carrier fluid is one of the problems of largest interest from the point of view of the technological applications of these systems. It can interfere with the magnetorheological response because of the non-uniform distribution of particles. Small particle size and high viscosity carrier fluids are recommended to reduce sedimentation but requirement of low viscosity in absence of magnetic field makes the problem more difficult as MRF

viscosity and yield strength are dependent on particle materials, particle concentration, particle size and size distribution, applied magnetic field, carrier fluids etc. [2-8]

Different approaches are attempted to control sedimentation:

- (a) adding thixotropic agents (carbon fibers, grease [9], silica nanoparticles[10])
- (b) adding surfactants (oleic [5], stearic acid)
- (c) adding magnetic nanoparticles [11]
- (d) the use of viscoplastic media as continuous phase [2]
- (e) water in oil emulsions as continuous phase. [8][10]

The most popular stabilizers are grease [9] and fumed silica [1][10]

One another category of magnetic fluid is ferrofluid (FF) which consists dispersion of Nano sized iron oxide particles. This reduces sedimentation problems because there is less density difference between particles and carrier fluid. The basic difference between FF (10nm) and MRF (in micron) is the size of the particles. [8][11] The particles are suspended by Brownian motion in FF. In usual magnetic field Brownian forces dominates the magnetic forces. [15] Therefore FF offers very less change in viscosity under very high magnetic fields compared to MRF. This makes their application limited where small viscosity change is required.[8] That is why FF is used in magnetic printers. In contrast, high magnetoviscous response of MRF makes them usable in wide technical application. [12]

The proposed work will focus on synthesis of MRF using Electrolytic Iron Powder, carrier fluids, and additives to reduce sedimentation rate.

## II. MATERIALS AND SYNTHESIS OF MAGNETORHEOLOGICAL FLUID

In this work Electrolytic Iron Powder (EIP), Silicon oil, Aerosil and Oleic acid are used as suspension, carrier fluid, stabilizer and surfactant. Electrolytic Iron powder is provided by Industrial Metal Powder, Pune and used without any further processing. EIP is characterized by shape and size distribution on Field Emission Scanning Electron Microscope (FESEM) and on Helos Particle Size Analysis Windox 5 respectively. Fig 1 shows SEM image of EIP which indicates needle shape of particles. The particle shape plays important role in yield stress of MRF. The needle shaped particles offer more surface area. Therefore offers more yield stress when they rub against rotating disk in rheometer.[13] Mean size of

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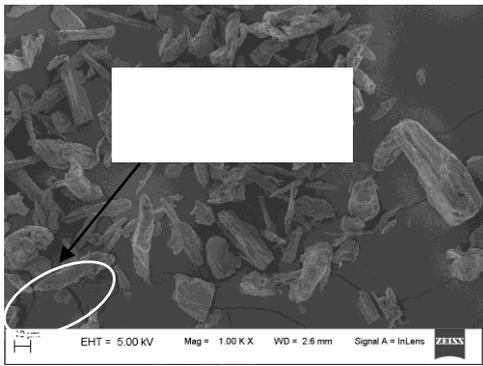


Fig. 1. Scanning electron microscope picture of Electrolytic Iron Powder with bar size 10 micron

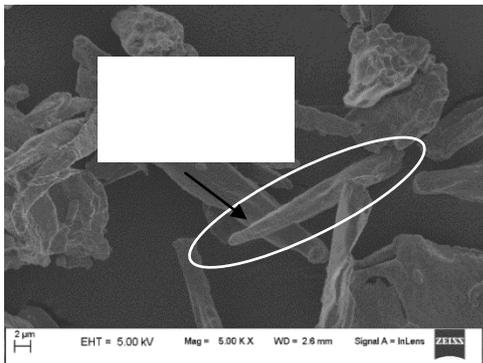


Fig. 2. Scanning electron microscope picture of Electrolytic Iron Powder with bar size 10 micron

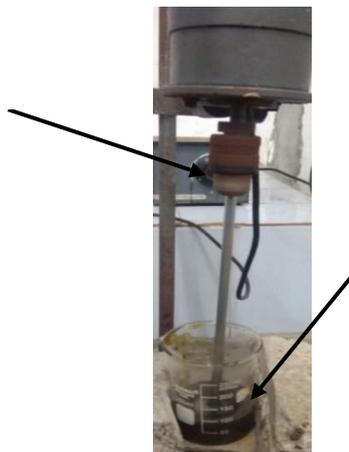


Fig. 3. Magnetorheological Fluid (MRF) synthesis.

Electrolytic Iron Powder is 20 micron from Helos Particle Size Analyzer. The silicon oil is used as carrier fluid having viscosity 300 cst and density 0.97 gm/ml. Aerosil 200 has density 0.084 gm/cm<sup>3</sup>. Oleic acid is used as surfactant which may also play some role in sedimentation as well. It has a density of 0.889 gm/ml.

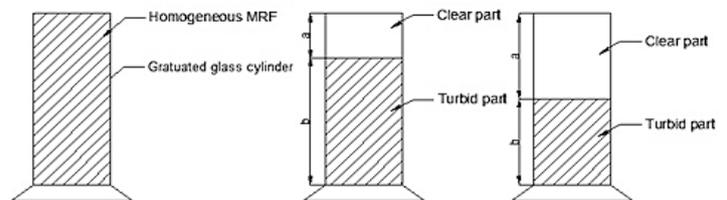
Synthesis of Magnetorheological Fluid (MRF) is performed using mechanical stirrer. Synthesis involves different variables which can play a major role in sedimentation of MRF. These variables (ingredient addition, stirring speed, stirring time etc.) are observed in sedimentation study of MRF in this article. MRF samples are synthesised by two different methods and their effect on sedimentation is also studied. In first method silicon oil, particles, oleic acid and aerosol are kept in glass beaker and stirred using mechanical stirrer. In second method all the

members of MRF are mixed stage by stage. The oleic acid, aerosil and particles are mixed in silicon in said order. The stirring set up is shown in fig. 3.

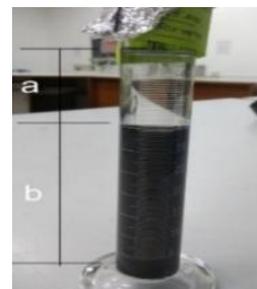
III. EXPERIMENTATION & METHODOLOGY

The major constituent of magnetorheological fluid (MRF) is carrier fluid. It works as medium for suspending magnetic particles and other additives. In present work silicon oil is used as carrier fluid which is colourless and oleic acid is also colourless. Therefore addition of this two constituent formulates colourless solution. Aerosil is white in colour hence addition of Aerosil produces white turbid solution. Later when Electrolytic Iron powder is added, it changes to dark grey colour. It is observed when MRF is allowed to settle then particle separation cause clear boundary separation between clear (absence of particles) and turbid part (containing particles) as shown in fig 4(a). Now movement of boundary separation is measured with time which gives sedimentation speed. In our work, borosil graduated cylinder is used for filling the MRF sample which has an internal diameter of 30 mm and capacity of 50 ml. The sedimentation rate is measured by taking ratio of clear part (a) and total sample height. [14] Sedimentation ratio (SR) can be given as:

$$SR = \frac{a}{(a+b)} \tag{1}$$



(a) Progressive sedimentation



(b) Settled sample

Fig. 4 Progressive sedimentation

IV. RESULTS AND DISCUSSION

This chapter contains effect of Aerosil, Oleic acid and some other synthesising parameters on stability of silicon oil based magnetorheological fluid.

A. Effect of particle concentration

General hypothesis when particle concentration is increased, it also increases particle density in sample, which

should also increase particle settling rate. However sedimentation rate reduces with increase in particle concentration as shown in fig 5. This may be due to increase in particle concentration as it causes large particles to settle rapidly but as particle size is not uniform as small particles are also present in the sample. Therefore settling of large particles causes upward flow of carrier fluid with high velocity. This fluid flow causes suspension of lighter density particles.[15]

**B. Effect of Aerosil 200**

Three different samples are prepared to study the effect of Aerosil as stabilizer. All the samples contain 20% EI and stirred for 2 hours at 1500 rpm. It is clear from the fig 6 percentage addition of Aerosil improves the stability of MRF.

**C. Effect of oleic acid**

Oleic acid is used as a surfactant. It is generally used to reduce particle agglomeration. However it can also change sedimentation behaviour of MRF. Therefore three sample are prepared with different concentration of Oleic Acid in absence of Aerosil with 2 hour stirring at 1500 rpm. Here our intuition turned to correct and oleic acid causes variation in stability of MRF as shown in fig. 7. This behaviour due to oleic acid might not be able covered surface of magnetic particles in initial 4 hours. Once particles are coated completely then agglomeration stops and stable MRF is achieved.[16]

**D. Combined effect of Oleic acid and Aerosil based MRF**

Only stabilizers are not sufficient to reduce abrasive wear, agglomeration and corrosion of particles surfactant are also to be used. Therefore to study combined effect, sample with both surfactant and stabilizer is prepared and its sedimentation observed. It is clear from fig. 8 due to presence of oleic acid with Aerosil stability of MRF reduces compared to MRF without Oleic Acid. But here we have to compromise in stability for long life of MRF.

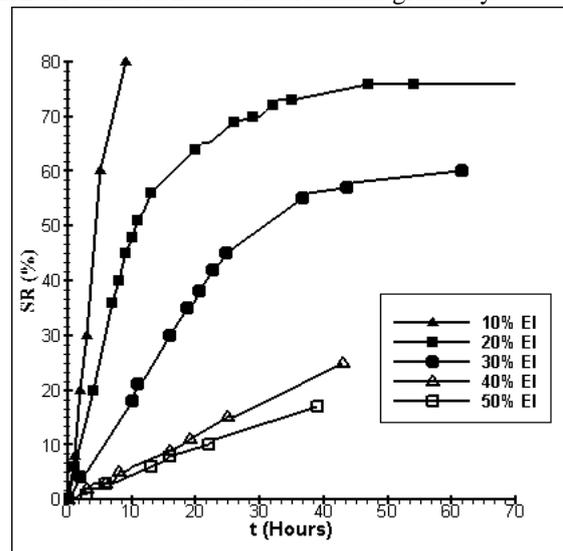
**E. Effect of stirring duration**

Stirring duration is important for homogeneous mixing of all the constituent of MRF. In literatures stirring hours are used above 12 hours but such a long duration reduces productivity of MRF. Therefore here we have prepared MRF with three different stirring duration. When compared to a 10 hours stirred sample against a 2 hours stirred sample, the initial result as well as the result after 6 hours seems to be similar, with slight variation in the middle. Hence in all our study, the sample were stirred for 2 hours only. The results of these are plotted in fig 9.

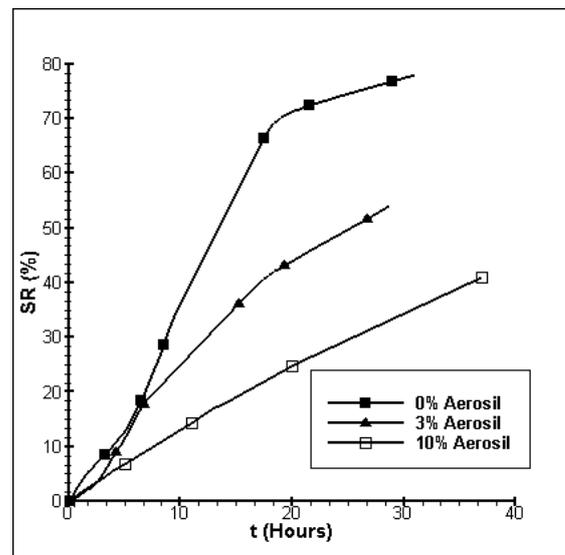
**F. Comparison of instantaneous and gradual addition of MRF constituent**

Two different approaches are used in synthesis of MRF. In first approach all the MRF constituents are added and stirred for two hours. In second approach in first half an hour oleic acid is mixed with silicon only, then in the next half an hour previously prepared solution is stirred with Aerosil and finally EI powder is stirred for one hour. Both these samples are observed and more stability is found for

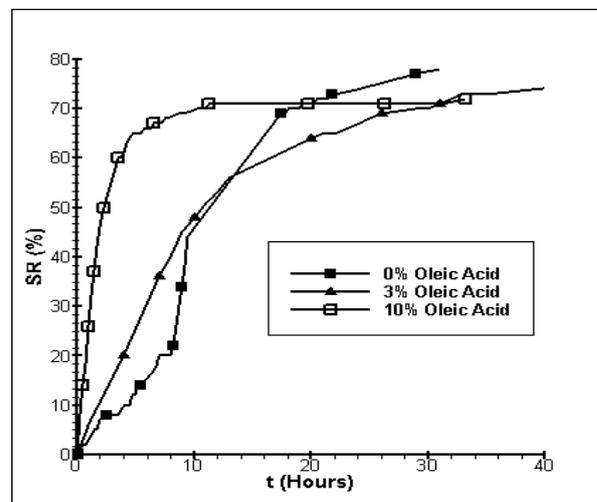
MRF in which constituents are added gradually.



**Fig. 5. Effect of particle concentration on sedimentation**



**Fig. 6. Effect of Aerosil on sedimentation**



**Fig. 7. Effect of Oleic Acid on sedimentation**



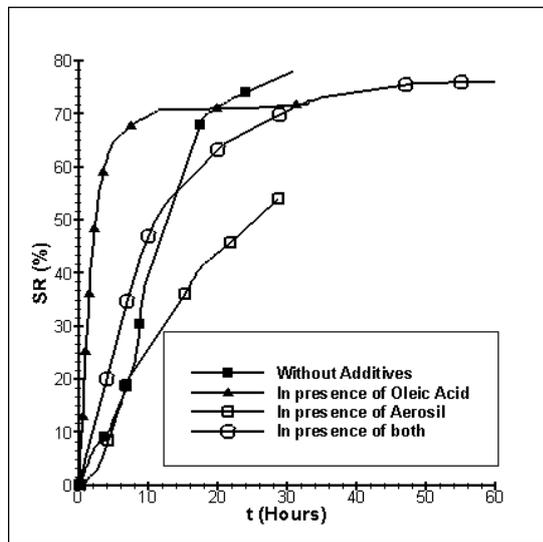


Fig. 8. Combined effect of Aerosil and Oleic Acid on sedimentation

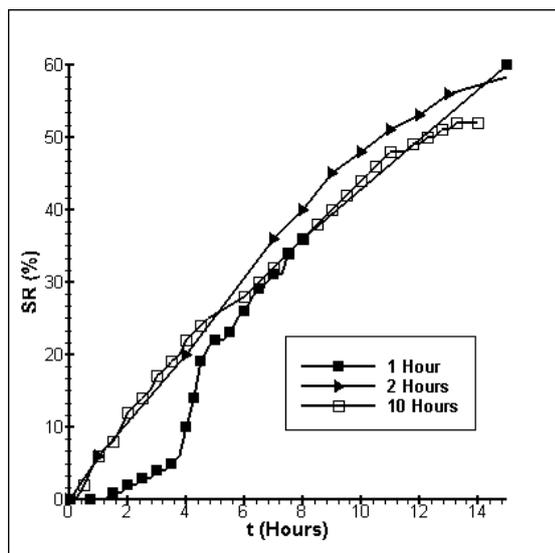


Fig. 9. Effect of stirring hours on sedimentation

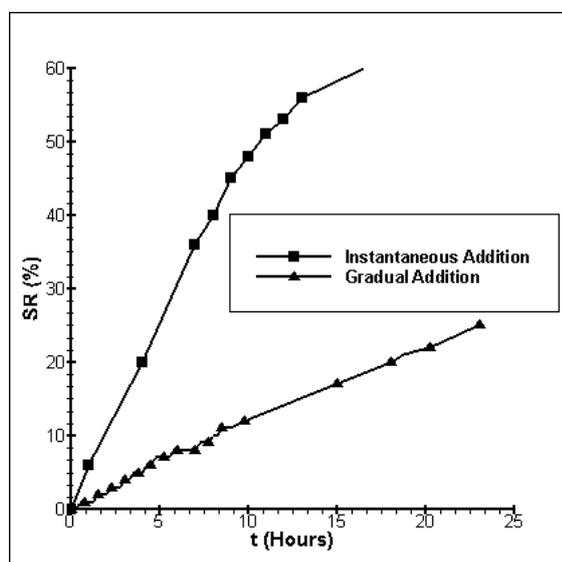


Fig. 10. Effect of synthesising method on sedimentation

## V. CONCLUSION

Additives are an important part of Magnetorheological Fluids (MRF). There is improvement in stability of MRF in presence of Aerosil but addition of surfactant (here oleic acid) causes reduction in stability but for long life of MRF, surfactants are an important constituent. Therefore some compromise has to be made in stability. During synthesis gradual addition of all the constituent gives remarkable improvement in stability. The stirring duration is also playing major role in the stability. Higher the stirring duration, higher is the stability. However this philosophy works well up to certain extent. There is no significant improvement in the stability after homogeneous suspension and complete particle coating of surfactant. Therefore 10 hours and 2 hours stirring are having marginal impact on the results. Sedimentation study based on visual observation is a qualitative way. Such method cannot give quantitative results. It is observed that some iron particles stick to the wall of measuring cylinder and that hinders the actual sedimentation at the interior portion. Therefore more accurate and reliable method has to be found out for quantitative study.

## VI. ACKNOWLEDGMENT

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