

Application of Copper doped Cadmium Sulphide Nanocrystallites for controlling Oral Pathogen activity

Sivanand.R, Chellammal.S, Manivannan.S

Abstract: In this study, we tend to incontestable the synthesis and antimicrobial action of Copper doped Cadmium Sulphide pathogens.Using averse oral Co-Precipitation method, Copper doped cadmium sulphide (Cu:CdS) nanomaterials are Synthesised. The sizes of the samples prepared are found by XRD (x-ray diffraction) method. SEM (scanning electron microscopy carried out to find the Morphological analysis. Materials present are confirmed by EDAX (energy dispersive analysis of x-ray spectroscopy) method. The outcomes showed that the proteins, which contain amine gatherings, played a diminishing and controlling obligation during the arrangement of Copper doped cadmium sulphide (Cu:CdS) nanomaterials in the colloidal arrangement. The antimicrobial action was evaluated against oral pathogens, for example, Bacillus subtilis, Micrococcus luteus, Staphylococcus aureus, Escherichia coli, Proteus vulgaris and Shigella flexneri and these outcomes affirmed that the Copper doped Cadmium sulphide nanomaterials are showing great bactericial action.

Keywords: XRD, EDAX, SEM, Antimicrobial activity, Oral pathogens

I. INTRODUCTION

Cadmium sulphide (CdS) nanomaterials have been used in vast number of applications in various fields; one of these in the medical field is dental care. Dental health is very important to maintaining a healthy mouth, teeth and gums. It will form healthy life. Oral pathogens will create many problems in human wellness like blood circulation and coronary diseases. Copper is very useful mineral for human health with bone, immune function, cardiovascular risk factor and cholesterol metabolism.

Most commonly affected diseases in human health is the dental problem, involve the commitment of microorganisms and development of biofilm on the common and reestablished tooth surface similarly.

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Here a biofilm is categorized as an aggregate of bacteria, from this cells stay in and on to an outside [1]. Nanostructured materials are an in fact huge article that has optical and electrical properties that depend on size of the nanomaterials. This is a direct result of detainment of the charge bearers in the constrained space of the nanomaterials [2, 3]. Semiconductor materials can have potential of antimicrobial applications with the Oral cavity [4]. As of late, II-VI semiconductor is playing consideration in tremendous fields because of their amazing and special properties when doped with other change metals [5]. Polymers additionally are great host materials as topping specialists and stabilizers since they hinder agglomeration and precipitation of the particles. Semiconductor nanomaterial used in biological labelling and photocatalysis. Copper is related with bone wellbeing, insusceptible capacity and expanded recurrence of contaminations, cardiovascular hazard and changes in cholesterol digestion [6]. Metals are utilized for a long time like gold, copper, silver, zinc, and titanium have pulled specifically consideration, each having different properties and spectra of movement. All oral nourishments, including toothpaste, presently incorporate powdered zinc citrate or acetic acid formation to control the improvement of dental plaque. Powdered titanium dioxide is utilized as a whitening specialist in toothpastes. The activities of antibacterial, antifungal and antiviral of sulphide nanomaterials have been examined in correlation with different metals. The utilization of silver nanomaterials has been all around area of medication applications, including inside the dental field, dental tar composites due to antibacterial effect. Expectation of caries and dentistry contamination is ordinarily focused at programmed or vague administration of the plaque bio film; bio films are a piece of our reality when brushing our teeth.

The bacterial activities are main parity to mechanical plaque control. Be that as it may, genuine times of presentation to antimicrobial researchers by practising tooth brushing and mouth expulsion might be blessing especially short, adding up to 30 seconds. In this work, we report extracellular combination of copper doped cadmium sulphide nanomaterials by the microscopic organisms, K.pneumoniae, as reducer and understand the compelling elements for amalgamation strategy.



II. MATERIALS AND METHODS

Cadmium chloride (CdCl₂) and Sodium Sulphide (Na₂S) were utilized for the combination of CdS nanomaterials and ethylene glycol was utilized as a capping agent. One mole of cadmium chloride solution was taken in one beaker with including 0.1 mole of capping agent and one mole solution of sodium sulphide (Na₂S) arrangement was taken in another beaker.Both measuring beakers arrangements independently blended persistent with a magnetic stirrer at 380 rpm at room temperature and including the two arrangements drop by drop again mixing with magnetic stirrer for 30 minutes. The shade of arrangement changes to light yellow. Subsequent to blending, the arrangement was centrifuged at 12000 rpm at room temperature. The Precipitated test was washed 3 times with methanol at last filtered and dried with vacuum dryer. For copper doped cadmium sulphide, 1mg of copper fillings included with 1 mole of cadmium chloride arrangement, rehashed the above readiness technique to get fine powder of arranged samples. Structural phase of the samples is determined by Powder XRD diffraction studies. Using Scherrer formula the grain sizes are calculated. SEM and EDAX techniques carried out for morphological analyses and can confirm the presented elements in the sample.

Well dissemination technique was utilized to examine the antibacterial movement of the blended CdS nanomaterial. All the dishes, reagents, and media are utilized to sanitize sterilise at for 20 minutes at 120°C. The pathogenic microorganisms Candida albicans, Staphylococcus aureus, Lactobacillus sp., and Streptococcus sp, were utilized as strains models. The unadulterated bacterial societies were sub refined on supplement juices medium. These strains were swabbed consistently onto the separate Muller Hinton agar plates utilizing sterile swabs. Well of 6 mm width was made on Muller Hinton plates utilizing gel cut. Utilizing a micropipette, the distinctive focus like 100 µL, 200 µL, and 300 µL of CdS nanomaterials arrangement was poured in the well on all plates and hatched for 24 hrs at 37°C. After brooding, the various degrees of zone of hindrance (ZOI) of microorganisms were estimated.

III. RESULTS AND DISCUSSION

A. Structural Characterization

The XRD spectra of the synthesized Copper doped cadmium sulphide Cu:CdS nanocrystallites are shown in fig. 1. The XRD spectra of Cu:CdS exhibit several diffraction peaks which are matched with the diffraction peaks of the standard JCPDS (PDF. NO. 890440)

XRD diffractometer (Rigaku X-ray diffraction analyzers from MiniFlex) measurement is used to calculate crystallites size of the Copper doped cadmium sulphide nanocrystallites. The average size are determined by the scherrer formula using the equation (1) given by

size
$$d = K\lambda/\beta cos\theta$$
(1)

Where K=0.9 which denotes shape factor for spherical, λ = 1.5406 nm CuK α radiation the incident wavelength, β denotes full-width at half maximum for the individual peak and bragg diffraction angle is θ . XRD patterns of prepared

samples in the range of 2θ = 300 to 700 are displayed in fig. 1. Peaks are found as the cubic structures of Copper doped cadmium sulphide are (111), (220) & (311) respectively.

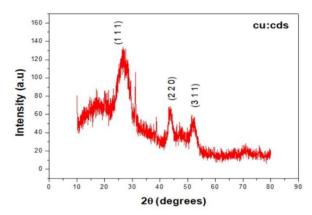


Fig. 1 shows XRD pattern of Cu:Cds nanocrystallites

The morphological analysis are carried by SEM measurement in Fig. 2 shows the shape of the nanocrystallites. Fig. 3. represents EDAX pattern of prepared samples which can be used to confirm the elements present in the sample.

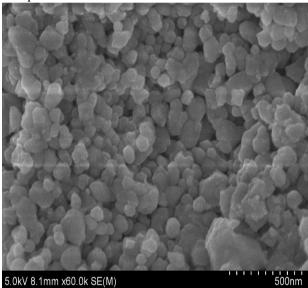


Fig. 2 shows SEM images of Copper doped cadmium sulphide

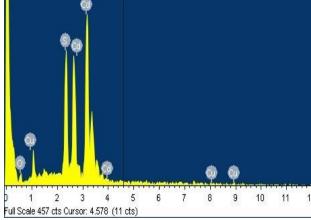


Fig. 3 shows EDAX pattern of Copper doped cadmium sulphide

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The calculated average size of the samples from the line broadening of the X-ray diffraction peaks using scherrer formula is 4 nm for Copper doped Cadmium sulphide nanocrystallites. Copper represents a very strong catalytic activity, a property that can be attributed to their large catalytic surface area.

B. Antibacterial activity Microbial strains

The microorganisms of Gram positive strains such as Staphylococcus aureus, Bacillus subtilis, and Micrococcus luteus as well as Gram negative strains such as Proteus vulgaris, Shigella flexneri and Escherichia coli were used for the evaluation of antibacterial activity.

Reference and control

Standard reference choosed here was Tetracycline. The controls comprise of cementing agar onto which was dissolvable, and the test mixes were solvent in it.

Aseptic conditions

A wooden box of $(1.3m \times 1.6m \times 0.6m)$ with a door is the aseptic chamber, is cleaned and irradiated by ethanol for cleaning and short wave UV light for irradiation.

Nutrient broth agar medium

Nutrient broth agar medium was prepared according to the standard methods (peptone-5 g, yeast-3 g, NaCl-5 g, distilled water- 1000 mL, agar-20 g) and was taken in 200 mL of distilled water in a 500 mL conical flask, made dissolved, stirred and finally autoclaved at 15 lbs and for 15 minutes at 121°C. The hot medium was poured in sterile petriplates which were kept in the aseptic laminar chamber. The medium was allowed to solidify for 15 min.

Agar well Diffusion method

Antibacterial activity of Copper doped Cadmium Sulphide nanoparticle was carried out using agar well diffusion method. The solidified nutrient agar in the petriplates was inoculated by dispensing the inoculum using sterilized cotton swabs which is previously immersed in the inoculum containing test tube and spread evenly onto the solidified agar medium. Five wells were created in each plate with the help of a sterile well-borer of 8 mm diameter. The Copper doped Cadmium Sulphide nanoparticle was then poured into each well containing 100, 150 and 200 μ g/mL concentrations. All the plates with nanoparticle loaded wells were incubated at 37°C for 24 h and the antibacterial activity was assessed by measuring the diameter of the inhibition zone formed around the well. Tetracycline (50 μ g) was used as positive control.

Table 1: Antimicrobial activity of Copper doped Cadmium Sulphide nanomaterials

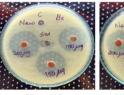
S. No	Organisms	Zone of inhibition mm			Standard
		100 µg	150 µg	200 μg	(Tetracycline)
1.	Bacillus subtilis	25	26	27	12
2.	Micrococcus luteus	24	25	26	11
3.	Staphylococcus aureus	27	28	30	12
4.	Escherichia coli	23	24	25	12
5.	Proteus vulgaris	24	26	28	12
6.	Shigella flexneri	14	15	17	15

Table 1 shows that the nanoparticle Copper doped

Cadmium Sulphide showed maximum zone of inhibition of 30 mm for Staphylococcus aureus, 28 mm for Proteus vulgaris, 27 mm for Bacillus subtilis, 26 mm for Micrococcus luteus, 25 mm for Escherichia coli and 17 mm for Shigella flexneri at 200 µg/mL concentration. The distinction in antibacterial action examined, was most likely related to cell capsular debasement and cell divider union hindrance by the nanoparticle just as fig. 4 shows that affectability of nanomaterials related with the distinctive cell divider structures of microscopic organisms, contains murein.

Nanomaterials discharge the particles, which react with the thiol packs in the proteins present on the bacterial cell film. Such proteins rock outcrop all through the microorganism cell surface awards helping supplements through the cell divider. The Amalgamation of CdS nanoparticle by using K. pneumoniae antimicrobial activity was tried against organisms Staphylococcus aureus (25 ± 0.33) and Lactobacillus sp. (23 ± 0.4) . From the results in table 1, we construed that the antibacterial activity is basically a result of CdS nanoparticle impregnated and included by the bacterial cell. As for instrument of antibacterial medicine development nanomaterials, gram negative minute creatures demonstrated more block zone than the gram positive organisms in light of the cell divider nature of the microorganisms. The gram positive bacterium course of action of the cytomembrane is assembled of significant layer of movie, including direct polysaccharide chains, and the gram negative minute life forms have the thin layer of layer.

In fig. 4 results shows that gram positive minute living beings have thick and artificially complex peptidoglycan in the telephone divider, so nanomaterials are not viably gone into the telephone. In any case, the gram negative bacterium have slight straightforward multilayered supermolecule materials inside the cytomembrane, so the nanomaterials did successfully go into bacterial cells and a while later exhibited a limitation zone higher than the gram positive microscopicorganisms.



Nano Still Copy Icopy



Bacillus

Micrococcu s luteus

Staphylococc us aureus

None © Ec Stal 200,49 100,49





Escherichi a coli

Proteus

Shigella

Fig. 4 Antimicrobial movement of Copper doped Cadmium Sulphide nanoparticle

The antibacterial effects and offset cadmium oxide nanomaterials can be helpful in the compelling contamination treatment brought living beings transmit a positive charge.



This makes an "electromagnet" interest between the microorganisms and treats the cell outside layer. The present examination shows that CdS nanomaterials have bactericidal activity against the entire test living thing.

Since this can be fundamentally open inside the nation and is in like manner used in center for helpful distinguishing strength administrator, the vivacious nanocompound from this can be masterminded and used effectively for evading the advancement of the oral pathogens.

Oral Pathogens can be controlled by Semiconductor Nanomaterials. The antiviral, antibacterial, and antifungal exercises of CdS blends have been expansively investigated with assessment of various metals. The usage of semiconductor nanomaterials has been all around thought of for a spread of therapeutic specialty applications, including, inside the dental field, an antibacterial root in dental gum composites. The usage of Cu:CdS nanomaterials, as a substitute for normal Cd, Cu, and S or complex Cu:CdS nanomaterials, has been portrayed and inspected with respect to their possible antimicrobial applications. Biosynthesis of nanomaterials is being inspected for a spread of potential applications, for example, ingestion to dental substitution materials and orthodontic pastes. The present work is exceptionally used underneath in vitro condition, in light of the fact that the biocompatibility of nanomaterials isn't completely tended to up until this point. Biofilm intensification is recognized to add to minor decay and as such the breakdown of sap based dental composites. Cu:Cds nanomaterials have encountered in vitro testing in biofilm culture attempt system. Cu:Cds nanomaterials intermixed into an arrangement of composites were appeared to through and through curb coccus sobrinus biofilm increase at centers. Hydroxyapatite material accessible in nanophase and nanocrystalline may be used as antimicrobial covering administrator to lessen the chance of microorganism movement.

IV. CONCLUSION

In this study, we reported the production of Copper doped Cadmium Sulphide nanomaterials by using Chemical Co-Precipitation method. The calculated average size of the samples from the line broadening of the X-ray diffraction peaks is 4 nm. Morphological analysis using SEM and the materials present are confirmed by EDAX. The utilization of semiconductor nanomaterials to sort out biofilm improvement encompassed by the oral depression, as an element of their bactericidal movement and conveyance capacities, is important and of genuine thought. This unpracticed science approaches is utilized to change biocompatibility of nanomaterials and extra supportive for growing innovation inquire about laborer to make the nanomaterials exploitation surface coatings and dental devices. Moreover, this procedure is eco-accommodating and nontoxic, and handlings of oral pathogens are likewise biocompatible.

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