# Better Utilisation of Bottom Ash in Coal Fired Thermal Power Station

J.Kenny Fernando, T.P. Meikandaan, M. Hemapriya

Abstract: In coal fired Thermal Power Station Bottom Ash is collected in Large quantities and stored in dykes. This project is done to convert the Bottom Ash in to a useful product which has good Adsorption Capacity of Hardness Causing Salts in hard water. This Bottom Ash is fused with NaOH and then Hydro Thermal synthesis was done and the extract was mixed with Sodium Aluminate and synthesis was done. The molarities of NaOH, temperature and duration was varied and synthesis was done. The synthesized product was tested for adsorption of Ca ions from Hard Water and found that it adsorbed the Ca ions.

Key Word: Sodium Hydroxide, Sodium Aluminate, Adsorption, Hydro Thermal Synthesis.

# I. INTRODUCTION

In coal fired Thermal Power Station in India lot of Ash is generated. Out of this Fly is finding lot of application and hence it is utilized in more quantity. But the Bottom Ash is not used and it is stored in dykes. The maintenance of the dykes is a great challenge and involves huge amount of money. There is a lot of environmental challenge due to leaching of toxic materials and breach of bunds. In order to prevent this problem this project is taken to convert the Bottom Ash in to useful product. The sample of Bottom Ash was taken from North Chennai Thermal Power Station for doing this project work.

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# II. MATERIALS AND METHODS

## A. Materials

The materials used for this experiment are 2 kg of bottom ash collected from the Ash Dyke of North Chennai Thermal Power Station. Sodium Chloride, Sodium Hydroxide Pellets, Calcium Chloride Fused , Magnesium Sulphate Heptahydrate, Sodium AluminateEquipments used were Muffle furnace, Air Oven , Electronic Weighing Balance, Magnetic stirrer with hot plate. Ceramic Crucibles.

## II. EXPERIMENTAL PROCEDURE

# A. Sample collection

2 kg of bottom ash collected from the Ash Dyke of North Chennai Thermal Power Station by using poclain in

polythene bags which were clean and dry. The samples were taken as two one kg bags.

# **B.** Sample Preparation

The sample was air dried and grinded to a powder form, it was then dried in a oven at 105 C for one hour.

# C. Extraction

The sample was tested for pH and it was found to be 9. Then the sample was mixed with Sodium Hydroxide and heated in a muffle furnace at 700 C for one hour.

The fused sample is mixed with D.M. water and heated to 60C with continuous stirring in a magnetic stirrer for one hour. Then the liquid was decanted and the sediments were removed. The decanted liquid was mixed with Sodium aluminate and stirred with magnetic stirrer. Then it was heated in a oven at 105C for 24 hours. The product obtained was washed with D.M water several times till the pH gets reduced to 8. Then it is dried in an oven at 105C for 2hrs.

# D. Testing

The extracted powder is tested for adsorption of hardness causing calcium salts and then the product is filtered. Calcium Chloride of 400 PPM is prepared and 50 ml of the

solution is taken and mixed with 0.5 gm of the synthesized powder and kept in amagnetic stirrer for 2



hrs . Then the solution is filtered in a filter paper the filtered liquid is tested for hardness and all the calcium was absorbed. were tried out. There are a lot of other possibilities which can be tried out by varying the Molarity of NaoH and the concentration of Sodium Aluminate .

III. RESULTS

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SL	Description	NCTPS	NCTPS
No		dyke	Inside
1	Insoluble Residue	93.57	93.67
	%		
2	Loss on Ignition	4.45	4.10
	%		
3	SO <sub>3</sub>	0.31	0.33
4	SiO <sub>2</sub>	60.38	60.17
5	Al <sub>2</sub> O <sub>3</sub>	26.59	27.04
6	Fe <sub>2</sub> O <sub>3</sub>	4.26	4.15
7	CaO	1.70	1.58
8	MgQ	1.01	1.09
9	Moisture	18	20
10	Blaine M2/Kg	84	60
11	90MIC	5.08	4.64
12	45 MIC	94.92	95.36
13	LR	0.2	0.25

## IV. RESULTS AND DISCUSSION

Use Coal Bottom Ash collected from Boiler Bottom was tested for pH and found to be slightly acidic . Then it was fused at 700 C for 1 hour and cooled and then tested for pH then it was found to be basic . Then the sample was fused without the removal of iron and with addition of  $0.1\ N$  of NaoH and tested for CEC which yielded very low CEC i.e. 2 meq/L.

Then the sample was fused with 0.1 N of NaOH with the removal of iron and tested for CEC which yielded a value of 20meq/L.

Then the sample was fused with sodium aluminate without the addition of NaOH and tested for CEC which yielded a value of 20 meg/L.

From the above tests the effect of removal of iron and addition of NaoH was understood. Moreover when the sample was dried and then ground a paste like substance was formed which was dried and then the various parameters like temperature and time were tried out. There are a lot of other possibilities which can be tried out by varying the Molarity of NaoH and the concentration of Sodium Aluminate .

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Ş1	Description	Value in %	
no			
1	SiO <sub>2</sub>	51.17	
2	Al <sub>2</sub> O <sub>3</sub>	38.7	
3	Fe <sub>2</sub> O <sub>3</sub>	8.99	
4	CaO	0.14	
5	MgO	0.14	

## V. CONCLUSION

Because From the above process it was concluded that fusion of the sample with Sodium Hydroxide followed by hydrothermal synthesis was optimum as per the facilities available within the limits of our capacity. The Si/Al ratio has a deciding factor. More over the hydro thermal synthesis has a very effective part. The synthesized product was very effective in adsorbing the hardness causing salts like Ca and Mg .Ca of 440ppm was completely absorbed in 0.5gms of the synthesized powder.

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